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### **Research Article**

### PHYSICO CHEMICAL ANALYSIS OF SHILAJATU (SHODHITA, MARITA AND SATWAPATHITA)

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### ABSTRACT

*Shilajatu* analysis is conducted as it is rare, controversial and herbo-mineral drug. Conventional literature reveals that only *Shodana* is sufficient enough and no need to go for further processes like Marana and Satwapatana. Unlike Rasaratna samuchhaya, the need for research on physical and chemical properties of *Shilajatu* is essential and unavoidable. Shilajatu is mineral pitch found in Himalaya rocks as exudations in the form of resin, hence the name Shilajatu. The works focuses on physical and chemical components present in four different forms of Shilajatu in the form of Raw, Shodhita, Marita, Satwapatita. The study is undertaken to assess organoleptic characters, pH, specific gravity, ash value, percentage of acid insoluble ash value and % of water soluble ash, moisture content, solubility in distilled water, and % of solubility. Now a day's stressful lifestyle, unhealthy diet, polluted environment, extremely pressurized work culture have created havoc among people affecting their physical and mental health leading to Medho vridhi vikaras (Hyperlipidemia). Shilajatu when used with proper Anupana can cure many diseases as said in Charaka chikitsa stana. Analytical study reveals presence of Elements like K, Ca, O, Cl, S, Na, Mg, Si, Al, Pb and oxides like K2O, CaO, SO3, MgO, Na2O, Al2O3, SiO3, P2O5, PbO, CuO, Cl in all the samples. Phytochemicals flavonoids, tannins, steroids triterpenoids, saponin, proteins were present in raw and Shodhita shilajatu.

### INTRODUCTION

*Shilajatu* is one among *Maharasa*, indicated for many diseases including *Sthoulya*, *Kusta*, *Prameha*, *Medogna*<sup>[1]</sup>. The *Shilajatu*, an exudation from the rock surface in the lower Himalayan hills used in ancient Ayurvedic system of medicine to treat obesity, diabetes, hyperlipidemia and several other ailments. The administrations of *Shilajatu* at proper time with proper *Anupana* cure almost all the curable diseases in the world<sup>[2]</sup>. It has properties like *Jwarahara* (antipyretic), *Pandughna* (cures anemia), *Pramehagna* (antioliabetu) and *Medachedaka* (which clears fat) etc., are mentioned in *Rasa ratna samuchaya* therefore it is used to treat hyperlipidemia.

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Hence to provide the scientific data, the study on physico chemical analysis of *Shodita*, *Marita* and *Satwapathita Shilajatu* and its effect on the serum lipids on albino rats fed with high fat diet are undertaken.

To analyze the physico chemical changes in Shodita Shilajatu, Marita Shilajatu and Satwapatita Shilajatu

### **MATERIALS AND METHODS**

To propose the methodology, following literature survey is carried to find the state-of- the-art methods in the related study specific to *Shodita*, *Marita* and *Satwapathita*, *Shilajatu* and anti-hyper lipidemic effect.

Screening of free radical scavenging activity and immunomodulatory effect of *Shilajatu* was proposed. *Shodita Shilajatu* was subjected for physicochemical analysis. Shiny black color, *Gomutra Gandhi* lump in appearance, pH was 480, specific gravity 0.997, ash value 17.4% w/w, 74% solubility in distilled water flavanoids, tannins triterpenoides, saponin, steroids, caratinoids is presented. Effect of *Shilajatu* on memory, anxiety and brain monoamines in rats is taken. The observed neurochemical effects are induced by *Shilajit*, indicating a decrease in rat brain 5-hydroxytryptamine turnover, associated with an increase in dopaminergic activity, helps to explain the observed non tropic and anxiolytic effects of the drug.

### Practical-1

### Shilajatu shodana

1) Raw Shilajitu- 1kg

2) Triphala kwatha- 4lt

3) *Goghrita* to smear on plate during drying the *Shodita Shilajatu*.

# Method of Preparation of *Shodhita Shilajitu Shlajatu Shodana*

### Procedure (1000gm)

*Shilajatu churna* was prepared in *Khalwa yantra*, powdered *Shilajatu* added in *Triphala kwatha* with constant stirring. Mixture was kept on stove, heated on *Mandagni* with constant stirring the mixture. When mixture attains syrup consistency it was poured on ghee smeared plate and kept for drying in sunlight. After complete drying *Shilajatu* removed from plate, powdered and stored in container.

### Table 1: Materials and Time Taken for the Shodana Process

Quantity of Shilajatu	Quantity of Triphala	Time taken to complete the process		
1 kg	4ltrs	3 hours		
Table 2: Weight of Chakrikas before Puta and weight of Basma after Puta				

Name of the <i>Puta</i>	Wt. of <i>Chakrikas</i>	Wt. of C.C.A ( <i>Vanyopala</i> )	Total weight of the <i>Basma</i> after each <i>Puta</i>
Kapotaputa	750 Grams	525 Grams	710 Grams

### Practical-2

Shilajatu Marana<sup>[3]</sup> Materials

- 1. Shodita Shilajitu- 200gm
- 2. Shodita Harthala- 200gm
- 3. Shodita Manashila- 200gm
- 4. Shodita Gandhaka- 200gm
- 5. Nimbu Swarasa- 400ml
- 6. Sarava- 3pairs

Above said drugs were taken one by one in collected the *Chakrik Kalwayantra* and mix thoroughly with spatula, stored in container. **Table 3: Materials Required for Giving Puta** 

*Nimbuswarasa* was added to mixture, *Chakrikas* were prepared and dried under shade.

Dried *Chakrikas* were placed in *Sharava samputa* and sealed by using *Gopichandana* such 3 *Sharava samputas* prepared, dried under shade later it was subjected to *Puta* separately by using 8 or12 cow dung cakes.

Sharavasamputa were allowed to Swanga sheeta after completion of Puta. Then carefully opened, collected the Chakrikas, powdered in Kalwayantra and stored in container.

Table 5. Materials Required for diving Fata						
No. of Puta	No. of <i>Sarava</i>	No. of No. of Cow dung caps Time taken		Time taken for		
(Kapota)	Samputa	Chakrikas	No. C.C	Wt. of	Puta	
1 <sup>st</sup> Puta	1	35	12	150 grams	20 min	
2 <sup>nd</sup> Puta	1	35	12	150 grams	20 min	
3 <sup>rd</sup> Puta	1	25	11	140 grams	20 min	

Table 4: Quantities of Vanyopala, number of Puta and weight of Bhasma after Marana

No. of Kapota puta	No. of <i>Chakrikas</i> in the <i>Sarava</i>	No. of Chakrikas	Wt. of C.C.A (Vanyopala)	Time taken for each <i>Puta</i>	Total weight of the <i>Basma</i> after each <i>Puta</i>
1 <sup>st</sup>	35	12	12 x 150 gm	20 min	260 grams
2 <sup>nd</sup>	35	12	12 x 150 gm	20 min	260 grams
3rd	25	11	11 x 150 gm	20 min	230 grams

Practical-3 Shilajithu satvapatana Materials (Drava Varga) Shudda Gunja- 100grams Shudda Madhu - 100grams

Shuddaguda- 100grams Shudda Grita- 100grams Shudda Tankana- 100grams Shudda Guggulu- 100grams Shudda Shilajitu- 100grams Nimbu Swarasa- 400ml **Procedure** 

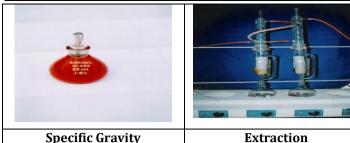
Above mention drugs were taken one by one in *Kalwayntra*. Mixed thoroughly, to this mixture *Nimbuswarasa* was added with continuous stirring till whole drug gets completely immersed in *Swarasa*. Triturating process was carried out continuously and constantly until mixture turns to semisolid pill mass

like state. After drying the mixture made into powder. The dried powder was kept in *Mushas* and its mouth sealed by lid using *Gopichandana*. *Musha* was placed on blower & *Teevragni* given for period of 40 min maintaining temp of 400°C. Next day morning the *Musha* was taken out carefully from blower and seal was opened. Solid mass of *Shilajatusatwa* collected from *Musha*, powdered and stored in container

<b>Table 5: Materials Re</b>	quired For Giving <i>Puta</i>

No's of Musha Qty before Satwapatan		Qty after Satwapatan	
1	270gm	170gm	
2 220gm		115gm	
Total wt before Satwapatan		Total wt after Satwapatan	
490gms		285gm	





# Physical and chemical analytical Study

### Samples of Shilajatu

Raw Shilajatu, Shodhita Shilajatu, Marita Shilajatu and Satwapatita Shilajatu

### A. Organoleptic characters

**B** Physico chemical analysis

### **C** Phyto-chemical analysis

### A. Organoleptic Characters of Shilajatu

Four samples of *Shilajitu* were subjected for organoleptic characters tests and results are given in table no: 06

### **B. Physico Chemical Analysis**

### 1) Solubility test

Solubility test of 04 samples of *Shilajatu* was conducted with the following solvents

ide

	-
i) Distilled water	ii) Ethyl alcohol
iii) Methanol	iv) Acetone.
v) Xylem.	vi) Toluene
vii) Benzene	viii) Carbon tetra chlor
ix) Petroleum ether	x) Chloroform

xi) Normal saline

### **Procedure (Solubility test)**

A pinch of *Shilajatu* was taken in a dry test tube with 1ml of solvent and shaken for 1 min, then observed for solubility, non-solubility and sparingly solubility. The results are noted and displayed in Table No: 07

Different solvents were used in different proportion and solubility was noted and displayed in Table No: 07

### 2) Percentage of Solubility

5gm of fine powder of *Shilajatu* was mixed with 100 ml of distilled water and stirred well. It was kept for 24 hours and filtered through Whatman filter paper no. 42. The filtrate was poured in china dish and then china dish was kept on water bath for drying. The filter paper along with residue was dried and weighed. The china dish was weighed along with sample. Percentage of solubility was calculated as per the standard formula.

 $W_1\mathchar`-$  Weight of crucible,  $W_2\mathchar`-$  Weight of solution after drying along with crucible:  $W_2\mathchar`-W_1/5\mathchar`-100$ 

i. Raw *Shilajatu* sample:  $W_{1=}45.75$ ,  $W_{2}=51.130$ 

% of solubility of raw *Shilajatu* sample= 51.130-45.75/5×100=17.6% in D.W

Same procedure was followed for *Shodhita*, *Marita* and *Satwapatita Shilajatu*.



ii. Percentage of solubility of *Shodhita Shilajatu* was 74% in D.W.<sup>[50,51]</sup>

iii. *Marita shilajitu* was 16.6% in D.W.

iv. Satwapatita Shilajatu was 15.8% in D.W.

### 3) Determination of pH

The pH meter was calibrated by using standard buffers of known pH 1.2, 4.0 and 9.2m. The reference electrode was thoroughly washed with distilled water every time and water was drained by using filter paper. 1% of *Shilajatu* solution (1 pH value of *Shodhita Shilajatu* 1gm of *Shilajatu* in 100ml of distilled water) was prepared. Then the tip of electrode was dipped and recorded.

1. pH value of raw *Shilajatu* sample= 4.6

2. pH value of *Shodhita Shilajatu* sample= 4.80

3. pH value of Marita Shilajatu sample= 5.6

4. pH value of *Satwapatita Shilajatu* sample= 8.5

# 4) Acidic and Alkaline Neutralizing Activity of *Shilajatu*

Three different pH Solution of having pH 1.2, 7.4 and 9.2 were prepared 1gm of *Shilajatu* was dissolved in 100ml of each of three pH media, and the change in pH of the solution was monitored of an interval of one hour from the time of addition of *Shilajatu* upto 12 hours, by using digital calibrated pH meter. The results were tabulated in Table No. 8 to 11.

### 5) Ash Value of Shilajatu

Air-dried powder of *Shilajatu* 5gm was taken in silica crucible incinerated in an electric burner till carbon free ash obtained. The percentage of total ash was calculated as follows.

Formula:

% of total ash =  $W_2$ - $W_1/5 \times 100$ 

Wt. of empty crucible (W<sub>1</sub>)

Wt. of crucible ash (W<sub>2</sub>)

Raw *Shilajatu* Sample: W<sub>1</sub>= 24.21, W<sub>2</sub>= 25.83

Ash value of raw *Shilajatu*= W2 – W1 (25.83-24.21)= 1.62

Ash value of raw *Shilajatu* = 1.62/5 ×100 = 32.5%.

Same procedure was followed for *Shodhita*, *Marita* and *Satwapatita Shilajatu* 

- *Shodhita Shilajatu* sample= 17.4%
- *Marita Shilajatu* sample= 9.4%
- Satwapatita Shilajatu= 4%.

### 6) Determination of Acid Insoluble Ash

Total ash obtained above was treated with 100ml of 6N dilute hydrochloric acid and boiled, insoluble matter was collected on ash less filter paper (Whatman No. 42). The residue was repeatedly washed with hot water, dried well, ignited in electric burner, cooled and weighed, the percentage of acid insoluble ash was calculated as follows -

[W<sub>1</sub>] Wt of crucible= 24.820gm

 $[W_2]$  Wt of the silica crucible + sample= 29.870gm

(W<sub>3</sub>)Wt. of Acid insoluble ash + crucible= 24.870gm

W3 – W1 = acid insoluble ash of *Shodhita Shilajatu*= 0.05 w/w

% of acid insoluble ash of Shodhita Shilajatu= 0.05X 100/5= 1%

Same procedure was followed for raw, *Marita* and *Satwapatita Shilajatu* 

ii. % of acid insoluble ash of raw *Shilajatu*.

iii. % of acid insoluble ash of *Maritata Shilajatu*.

iv. % of acid insoluble ash of *Satwapatita Shilajatu*.

Preparation of 6N HCL with 100ml distilled water,  $N_1V_1\text{=}~N_2V_2$ 

Where N1= Conversion factor (For HCL it is 36.5)

V1= Volume of HCL needed

N2= Required dilution

V2= Volume of distilled water used.

36.5 \* V1= 6 \* 100

V1= 6 \* 100/36.5 = 16.44ml

16.44ml HCL is needed to prepare 6N HCL with 100ml distilled water.

% of acid insoluble ash =  $W_2$ - $W_1/5 * 100$ 

### 7) Determination of water insoluble ash

Total ash obtained from procedure E was boiled with 25ml of distilled water and filtered through ash less filter paper (Whatman No. 42). The residue was washed with hot water and dried well, ignited in electric burner cooled and weighed the percentage of water insoluble ash

[W<sub>1</sub>] Wt of empty crucible= 24.820gm

[W<sub>2</sub>] Wt of the silica crucible + sample= 29.870gm

 $[W_3]$  Wt. of water insoluble ash crucible= 25.500gm

Wt of water insoluble ash (W3-W1): 0.68gm

i. % of water insoluble as h of Shodhita sample– 0.68 X 10015= 13.6%

Same procedure was followed for raw, *Marita* and *Satwapatita Shilajatu* 

ii. % of water insoluble ash of raw sample= 6.6%

iii. % of water insoluble ash of *Marita* sample= 5.8%

iv. % of water insoluble as h of Satwapatita sample= 13.84%

## 8) Determination of Specific Gravity

Empty specific gravity bottle was weighed, bottle filled with distilled water was again weighed same bottle was then filled with 1% of sample (100ml distilled water+1gm of *Shodita Shilajatu*) was weighed.

The specific gravity of the sample was calculated as,

 $W_3 - W_1 / W_2 - W_1$ 

Where,  $W_1 = Wt$  of empty bottle

W<sub>2</sub> = Wt of bottle filled with distilled water

 $W_3$  = Wt of bottle filled with 1% sample solution.

Raw *Shilajatu* Sample:  $W_1$ =19.290,  $W_2$ =42.080,  $W_3$ =42.390

Sp gr= 42.390-19.290/42.080-19.290=23.10/ 22.79=1.0136

Specific gravity of raw Shilajatu=1.0136

Same procedure was followed for raw, *Marita* and *Satwapatita Shilajatu* 

specific gravity of *Shodhita Shilajatu*= 0.9973

Specific gravity of Marita Shilajatu= 1.00834

Specific gravity of Satwapatita Shilajatu = 1.0136.

### 9) Determination of Moisture Content

A clean and dry petri dish was weighed 5gm of dried sample powder was taken in the petri dish and the petri dish was kept in hot air oven and weighed. Again petri dish was kept in hot air oven for one more hour at 105°C The procedure was repeated until the constant weight of petri dish containing the powder of sample was obtained % of moisture content= Difference of weight/Wt of sample ×100

 $= W_2 - W_3 / W_1 \times 100$ 

Initial weight of Petri dish= 81.4gm

W1 = Initial wt of the sample = 5gm

 $W_2$  = Initial wt of petri dish with sample = 86.4gm

 $W_3$  = Final wt of petri dish with sample (after drying). =86.3

Moisture content raw *Shilajatu* Sample= 86.400-86.300/5×100=2%

% of moisture content of raw *Shilajatu*= 2%

% of moisture content of *Shodhita Shilajatu*= 1.4%

% of moisture content of *Marita Shilajatu*= 0.2%

% of moisture content of *Satwapatita Shilajatu*= 0.2%

Phyto-Chemical Analysis [24, 25, 26]

All the 4 samples of *Shilajatu* were subjected for phytochemical analysis

### 1. Detection of Alkaloids

Test drug sample solution+Wagner's reagent (kI solution). Reddish brown colored ppt was not observed in test drug sample. It indicates the absent of alkaloids

### 2. Detection of Carbohydrates

Benedict's test: (test for reducing sugars)

5ml of Benedict's reagent was heated in a test tube. 8-10 drops of test drug sample solution was added. Brick red ppt was not observed in sample. It indicates the absence of carbohydrate.

### 3. Detection of Tannins

Test drug sample solution +FeCl<sub>3</sub> blue colour ppt was observed in test drug sample. It indicates the presence of tannins.

### 4. Detection of Steroids

Salkowskls test, test drug sample Solution + concentrated H<sub>2</sub>SO<sub>4</sub>+ chloroform, blood red ppt was observed in the tested drug sample. It indicates the presence of steroids.

### 5. Detection of Triterpenoids

Salkowski's test. test drug sample solution concentrated  $H_2SO_4$  + chloroform + water (10ml choroform 90ml water i.e., 1:9) yellow ppt was observed in a sample. It indicates the presence of triterpenoids.

### Lemental analysis graphs by XRF

### **Raw Shilajatu**

### 6. Detection of Flavonoids

Sodium Hydroxide solution test: Test drug sample Solution + 10% NaOH solution, vellow ppt was observed in sample. It indicates the presence of flavonoids.

### 7. Detection of Proteins

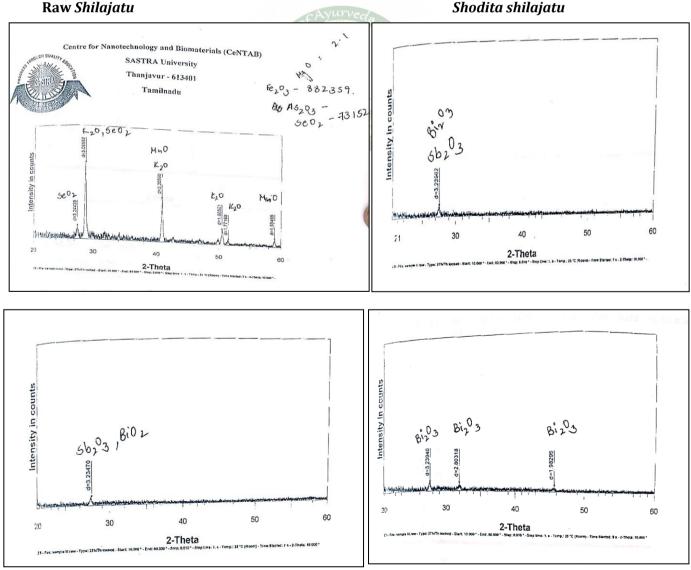
Million's Biurct test: Test drug sample Solution + 2ml 10% NaOH solution + 2m110% CuSO<sub>4</sub>. Blue colour ppt was observed in sample. It indicates the presence of proteins.

### **Detection of Saponins (Foam Index Test)**

1 ml extract of test drug solution + 20ml distilled water (shaked for 15 minutes), foam was observed in a sample. It indicates the presence of saponins

### HPTLC, XRD and Elemental Analysis by XRF

Four samples of Shilajatu: Raw Shilajatu, Shodita Shilajatu. Marita Shilajatu, Satwapatita Shilajatu are subjected for the above mentioned analysis. XRD graphs of raw, Shodhita, Marita, Satwapatita Shilajatu are shown below.



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	Table 6: Organoleptic Characteristics of Samples of Shilajatu					
S. No	Organoleptic Characters	Raw Shilajatu	Shodhita Shilajatu	Marita Shilajatu	Satwapatita Shilajatu	
1	Colour	Shiny black	Shiny Black	Shiny Black	Black,	
2	Odour	Gomutra Gandi	Gomutra Gandhi	Odourless	Odourless	
3	Taste	Astringent	Astringent	Tasteless	Tasteless	
4	Touch	Smooth	Smooth	Smooth	Smooth	
5	Appearance	Lump	Lump	Powder form	Powder form	

Table 7: Physical Constants of 04 Samples of Shilajatu

S. No	Physical Constants	Raw	Shodita	Marita	Satwapatita
1	рН	4.6	4.80	5.6	8.5
2	Specific gravity	1.014	0.997	1.0083	0.998
3	Ash value	32.5%	17.4%	9.4%	4%
4	% of acid insoluble ash	1.8%	1%	1.5%	2.3%
5	Water insoluble ash	6.6%	13.6%	5.8%	13.84%
6	Moisture content	2%	1.4%	0.2%	0.2%
7	Solubility in DW	S	S	NS	S
8	% of Solubility	17.4%	74%	NS	15.8%

 Table 8: Shows Phytochemical Constituents of Samples of Shilajatu

Phytochemical constituents	Raw Sample	Shodhita sample
Alkaloid	-ve	-ve
Carbohydrates	-ve	-ve
Flavonoids	+ve	+ve
Tannins	+ve	+ve
Triterpenoids	JAPR +ve	+ve
Steroids	+ve	+ve
Proteins	+ve	+ve
Saponins	+ve	+ve

**Table 9: Percentage of Elements in all the Four Samples** 

Elements	Sample-I	Sample-II	Sample –III	Sample – IV
0	4.43 %	8.07%	11.6%	9.17%
К	0.442%	0.468%	1.65%	0.576%
Са	0.649%	1.08%	1.14%	0.870%
Cl	0.660%	0.725%	2.48%	0.736%
S	0.602%	0.776%	0.183%	0.760%
Na	1.82%	2.08%	3.52%	0.733%
Mg	1.08%	2.18%	1.46%	
Fe	0.966%	1.25%		0.835%
Si	2.22%	2.88%	2.20%	1.69%
Hg	3.15%	-	-	-
Al	4.67%	5.14%	4.40%	3.89%
Pb	8.22%	2.33%	9.69%	6.31%

Р	6.16%		5.74%	2.18%
Pd	12.6%			
Mn	10.2%		-	10.9%
Sr	3.97%	-		
Cr	18.5%	-	-	-
Мо	14.4%			
Cu	12.8%		3.35%	3.04%
Zn		12.5%		
Rb			0.0565%	
As			0.0565%	0.364%
Ti				9.40%

Table 10: Percentage of Oxides in all the Four Samples

Oxides	Sample-I	Sample-II	Sample –III	Sample – IV
K <sub>2</sub> O	0.442%	0.468%	1.65%	0.576%
CaO	0.649%	1.08%	1.14%	0.870%
SO <sub>3</sub>	0.602%	0.776%	0.183%	0.760%
MgO	1.08%	2.18%	1.46%	1.80%
Na <sub>2</sub> O	1.81%	2.08%	3.52%	0.733
Fe <sub>2</sub> O <sub>3</sub>	0.966%	of http://ijapr.in	1.61%	0.835%
SiO <sub>2</sub>	2.22%	2.88%	2.20%	1.69%
Al <sub>2</sub> O <sub>3</sub>	4.67%	5.14%	<mark>4</mark> .40%	3.89%
$P_2O_5$	6.16%	2.33%	5.74%	2.18%
PbO	8.22%	5.14%	9.69%	6.31%
MnO	10.2%	16.2%	42135	10.9%
MoO <sub>3</sub>	14.4%	JAN		
$Cr_2O_3$	18.5%			
SrO	3.97%			
CuO	12.8%	15.7%	3.35%	3.04%
ZnO		12.5%		
Rb <sub>2</sub> O		14.7%		
$AS_2O_3$			0.0565%	0.364%
TiO <sub>2</sub>				9.40%
Cl	0.660%	0.725%	2.48%	0.736%

### **RESULTS AND DISCUSSION**

Raw, *Shodhita, Marita* and *Satwapatita Shilajatu* all these four samples were subjected to organoleptic, physical constants, chemical, HPTLC, XRD and elemental analysis. (XRF)

*Gomutra Gandhi* in odour, smooth in touch, astringent in taste, appearance in lump form and was black colour observed in raw and *Shodhita Shilajatu*. Shiny black in colour was observed in *Marita Shilajatu*. Black colour was observed in *Satwapatita Shilajatu*. Odourless, tasteless, smooth in touch, powder form in

appearance was observed in *Marita Shilajatu* and *Satwapatita Shilajatu*.

Raw *Shilajatu* was subjected for physical constants specific gravity of raw *Shilajatu* was= 1.0136 *Shodhita Shilajatu*= 0.9973, specific gravity of *Marita Shilajatu*= 1.00834, specific gravity of *Satwapatita Shilajatu*= 1.0136.

All the four samples (raw *Shilajatu, Shodhita Shilajatu, Marita Shilajatu* and *Satwapatita Shilajatu*) were subjected for solubility test by using different solvents. All the three samples raw, *Shodhita*,

*Satwapatita* were soluble only in distilled water and were not soluble in any of the solvent but *Marita Shilajatu* was not soluble in any of the solvents.

**Phytochemical Analysis:** Raw and *Shodhita Shilajatu* was subjected to phytochemical analysis and showed presence of flavanoids, proteins, tannins, triterpenoids, saponins, steroids.

**Element analysis** of four samples (raw, *Shodita*, *Marita*, *Satwapatita Shilajatu*): Elements like K, Ca, O, Cl, S, Na, Mg, Si, Al, Pb and oxides like K2O, Cao, SO3, MgO, Na2O, Al2O3, SiO3, P2O5, PbO, CuO, Cl were noted in all the samples. Pd, Sr, Cr, MO were only noted in raw *Shilajatu*. Zn was noted in sample II, Cu noted in all samples except II sample

**Sample I (Raw** *Shilajatu***):** Pd, Hg, Cr, Mo, Sr noted only in this sample.

**Sample II (Shodita Shilajatu):** This sample which was subjected to Shodhana with Triphala kwatha: Zn (12.5%), Rb (14.7%) presence of these two noted and Pd, Hg, Cr, Mo, Sr were disappeared.

**Sample III** (*Marita Shilajatu*): This sample is obtained from incineration process with addition of arsenic and sulphur compounds to obtain fine powder. The analysis shows the presence of Cu, arsenic as As<sub>2</sub>O<sub>3</sub> (0.0565) (arsenic oxide) and SO<sub>3</sub> (0.183) sulphur trioxide. But Pd, Rb, Cr, Sr, Hg, which were noted in raw sample, absent in this sample.

**Sample IV (Satwapatita Shilajatu):** The Shodhita Shilajatu sample was mixed with honey, jaggery *Guggulu*, ghee, borax powder and subjected for intense heat to get extract. This sample contains an extra element Cu, titanium (Ti) (9.40%) and its oxide TiO<sub>2</sub> (9.40%). But Pd, Rb, Cr, Sr, Hg, which were noted in raw sample, absent in this sample.

**XRD reports of 4 Samples**: In XRD analysis of four samples (Raw *Shilajatu, Shodita Shilajatu, Marita Shilajatu, Satwapatita Shilajatu*) spherical shape were detected in all the four samples (raw, *Shodhita, Marita* & *Satwapatita Shilajatu*).

**Raw** *Shilajatu*: 4.49μm minimum size and 11.08μm maximum size of particle was noted.

**Shodhita Shilajatu:** 3.20μm minimum size and 21.00μm maximum size of particle was *Marita Shilajatu*: 18.59μm minimum size and 64.80 μm maximum size of particle was noted. *Satwapatita Shilajatu*: 5.11μm minimum size & 19.07μm maximum size of particle was noted.

**Sample I (Raw Shilajatu):** Elements Hg, Pd, Sr, Cr, Mo and oxides  $MoO_3$  and  $Cr_{2O_3}$  are significantly noted only in this sample but absent in other three samples

**Sample II** (*Shodita Shilajatu*): Elements Rb, Zn and its oxides  $Rb_20$  and Zno are found only in this sample but absent in other three samples.  $Al_2O_3$ ,  $SiO_2$ , MgO increased than in other sample.

**Sample III (***Marita Shilajatu***):** Elements P, Al, Pb and oxides K<sub>2</sub>O, Na<sub>2</sub>O, Fe<sub>2</sub>O<sub>3</sub>, are increased only in this sample but absent in other two samples.

**Sample IV** (*Satwapatita Shilajatu*): Elements Al, Pb, Mn and oxides  $Tio_2$ , and  $As_2O_3$  are increased only in this sample than other samples.

Elements Al & Pb are noted in all the 4 samples (Raw Shilajatu, Shodita Shilajatu, Marita Shilajatu, Satwapatita Shilajatu), whereas Hg, Pd, Sr, Cr, Mo are found in only raw Shilajatu but absent in other 3 samples. Mn is significantly increased in Satwapatita Shilajatu Rb & Zn have significantly increased in Shodita Shilajatu. P was found in sample-1 raw Shilajatu which was decreased in Marita Shilajatu and Satwapatita Shilajatu.

Na, Mg, Si have slightly increased in all the sample i.e., (*Shodita Shilajatu*, *Marita Shilajatu* & *Satwapatita Shilajat*) but Sr is slightly found in only Raw K, Ca, Cl, S, Fe are present in traces in all the samples.

Oxides PbO & MnO are noted in major percentage in all the samples except III sample MoO & Cr<sub>2</sub>O are found in raw *Shilajatu* but absent in all the other 3 samples, where as Zno & Rb<sub>2</sub>O are significantly noted in *Shodita Shilajatu* but absent in all the other 3 samples. Tio<sub>2</sub> is found in *Satwapatita Shilajatu* but absent in all the other 3 samples.

Oxides Mgo, Sio<sub>2</sub> &  $Al_2O_3$  are noted in all the samples.  $k_2O$ , Na<sub>2</sub>O & Fe<sub>2</sub>O<sub>3</sub> are increased in only *Marita Shilajatu* but found in traces in all the other samples but SrO is noted in only raw *Shilajatu* but absent in all the other 3 samples)

### Four samples were subjected to HPTLC

**Sample 1 (Raw Shilajatu):** As the raw sample contains water, nitrogenous matter, calcium and mica. The HPTCL shows various peaks might be of various nitrogenous compounds. The calcium and aluminium oxide might have formed some complexes with nitrogenous matter. This may be assumed from XRD & XRF reports.

**Sample 2 (Shodita Shilajatu)** The HPTLC of purified sample that is purified by *Agnitapi* process, *Shilajatu* was subjected for *Shodana* with *Triphala kwatha* where nitrogenous matter might have evaporated in the form of different gases remaining oxides of calcium, aluminium in the residue it may be considered from the XRF and XRD reports. Rf values and area % of the peak reveals the components of fulvic acid, bioflavonoids and taroid complex, the *Amalaki* peak with maximum rf value 0.18 area % 2.38 and *Bibitaki* peak with maximum rf value 0.45 area % 13.35 were also noted. HPTLC of sample 3 preparation from purified sample II is mixed with *Shodhita Harathala, Manashila* and *Gandhaka*. During the process the sulphur content of the above three drugs evaporated

mostly in the form of sulphur oxide (So2) and the residue *Bhasma* contains mostly oxides of calcium, magnesium, iron, copper, zinc and some other oxides of arsenic. This may be assumed from the XRD & XRF reports

**Sample 4** (*Satwapatita Shilajatu*): The *Shodhita Shilajatu* was mixed with *Guda, Gunja, Guggulu, Godugda, Ghrita* and *Tankana* and subjected for *Satwapatana* (heating the mixture upto 400°C).

During this process most of the carbohydrates, fatty material were evaporated in the form of  $CO_2$ , moisture etc. The residue mineral (*Shilajatu*) is rich in oxides of calcium, iron, sodium, boron etc this also may be considered from XRF and XRD reports.

From qualitative and quantitative of components like elements, phyto-chemical constituents, preparation and economic, therapeutic activity point of view *Shodhita Shilajatu* considered better.

### CONCLUSION

As the raw sample contains water, nitrogenous matter, calcium and mica, the HPTCL shows various peaks might be of various nitrogenous compounds. *Shodita Shilajatu* components of oxides of calcium, aluminium remains in the residue of purified sample. *Marita Shilajatu* residue *Bhasma* contains mostly oxides of calcium, magnesium, iron, copper, zinc and some other oxides of arsenic. The mineral (*Shilajatu*) is rich in oxides of calcium, iron, sodium, boron.

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